

Polyatomic Compounds - Examples

Give the name and formula for the compound formed by:

1) Sodium and Sulfate

$\text{Na}^+$   $\text{SO}_4^{2-}$   $\text{Na}_2\text{SO}_4$  Sodium Sulfate

2) Potassium and Carbonate

$\text{K}^+$   $\text{CO}_3^{2-}$   $\text{K}_2\text{CO}_3$  Potassium Carbonate

3) Ammonium and Nitrate

$\text{NH}_4^+$   $\text{NO}_3^-$   $\text{NH}_4\text{NO}_3$  Ammonium Nitrate

4) Beryllium and Nitrate

$\text{Be}^{+2}$   $\text{NO}_3^-$   $\text{Be}(\text{NO}_3)_2$  Beryllium Nitrate

Apr 1-9:37 AM